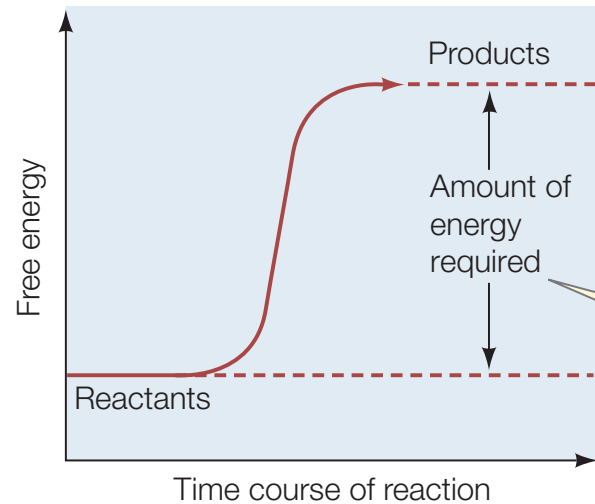
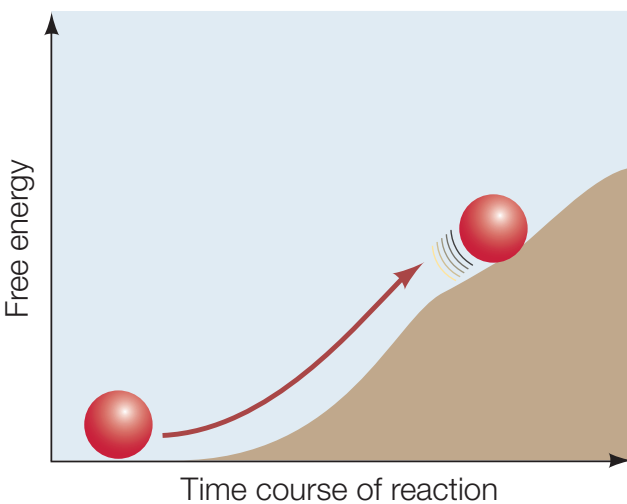
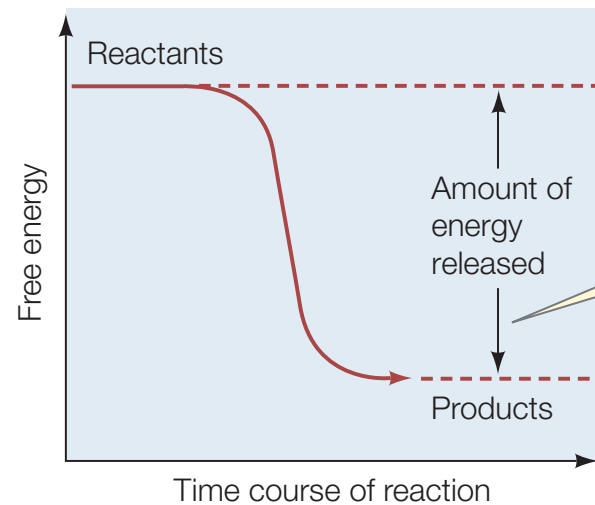
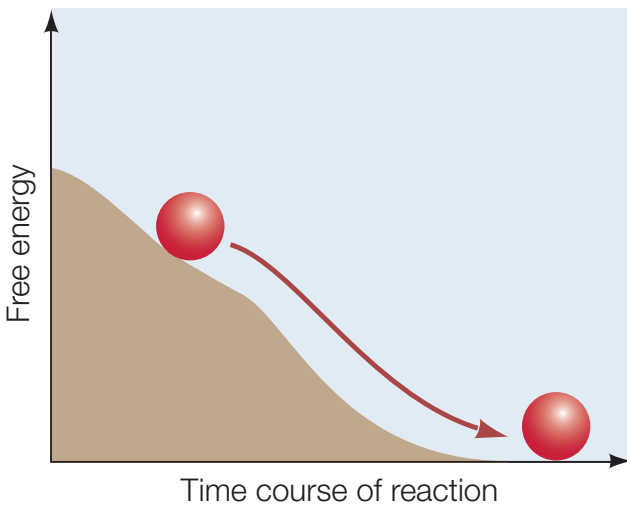


(A) Endergonic reaction



*Energy must be added for an endergonic reaction, in which reactants are converted to products with a higher energy level.*

(B) Exergonic reaction



*In an exergonic reaction, energy is released as the reactants form lower-energy products.*